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The Role of
Geochemistry in
Gineral SystemsBY:Carl Brauhart
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Setting out to be Creative

"If you want to end up somewhere different, you need to start somewhere different"

Brian Eno Music Producer



Take-Home Messages

In Mineral Exploration there are THREE main things that whole-rock geochemistry can help us with

- 1. Lithology (Mostly Immobile Element Geochemistry)
- 2. Alteration (All About Mineralogy)
- 3. Metal Signatures (Direct Detection of Mineralisation With Multielement Geochemistry)





Mineral Systems = Context







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Immobile Element Geochemistry



Immobile Element <u>RATIOS</u> Define Rock Types





- Immobile elements neither enter, nor leave a rock mass during alteration or weathering
- Concentrations may change, ratios remain constant
- Key elements include Th, Nb, REE, Zr, Ti and Sc

Immobile Elements

2

3

5

6

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He

Ne

Ar

Kr

Xe

Rn

Og

F

CI

Br

At

Ts

Immobile-Incompatible Element Classification



Incompatible element pairs maintain very similar ratios across a wide range of compositions

That makes them very useful for discriminating different <u>magma</u> <u>series</u>

Lithogeochem Calculator



Lithogeochem **Calculator** compares 13 element ratios to quantify difference between profiles for two samples Compatible elements: are avoided because they vary according to

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Discriminating Magma Series

Use incompatible element ratios to discriminate between magma series Use compatible elements to discriminate within a magma series



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Lithogeochem Calculator

Panorama VHMS District

440 rock chip samples classified using Lithogeochem Calculator

Spatially coherent domains result





Panorama VHMS: Rapidly Classify Bi-plots



Panorama VHMS: Upper and Lower Volcanic Suites

- Lower Suite
- Upper Suite
- Under Sulphur Springs
- Calc-alkaline Rhyodacite
- Granophyre-Rhyolite
- 😑 Outer Phase Granite
- Microgranite
- 😑 Inner Phase Granite
- Late Dolerite



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Th Nb Ta La Ce Pr Nd Zr $\,$ P $\,$ Sm Eu Ti Gd Tb Dy $\,$ Y Ho Er Yb Lu V $\,$ Sc $\,$



Previously Unrecognised Suite

- Lower Suite
- Upper Suite
- Under Sulphur Springs
- Calc-alkaline Rhyodacite
- Granophyre-Rhyolite
- 😑 Outer Phase Granite
- Microgranite
- 😑 Inner Phase Granite
- Late Dolerite

Volcanic and granite

Third suite has subtly higher Th/Yb & La/Yb



Panorama VHMS: Outer and Inner Phase Granite

- Lower Suite
- Upper Suite
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Panorama VHMS: Rapidly Classify Bi-plots



How Has This Helped?

- 🔵 Lower Suite
- Upper Suite
- Under Sulphur Springs
- Calc-alkaline Rhyodacite
- Granophyre-Rhyolite
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- Ti-Zr has been used to validate mapping of compositions basalt through to rhyolite
- Detailed immobile element geochemistry defines a break in volcanic stratigraphy – VHMS implications
- Four major magma series helps unravel the order of events in the mineral system





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2. Alteration Geochemistry



Alteration Diagrams

On any diagram, ask "What minerals are likely to be driving trends on my diagram?". <u>It's all about minerals</u>



Alteration Elements



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Panorama VHMS Mineral System





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Panorama VHMS Mineral System





Panorama VHMS Mineral System: Mass Transfer Maps



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How Has This Helped?







- Na/Al versus K/Al molar ratio plot confirms alteration mapping
- Architecture of alteration map can be interpreted as a convective hydrothermal system: discharge zones are targets
- Albite alteration coincides with zone of strong metal leaching = high temperature reaction zone

Choose Diagrams Appropriate to Your Mineral System

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3. Mineralisation Signatures



Principal Component Analysis

- PCA is very useful to identify multielement associations: Mineralisation
- Rather than 40, or 60 individual elements, a handful of ranked scaled eigenvectors
- The proportion of variation owing to each element association (process) is defined <u>Single element maps mix all these processes together</u>



PCA Step 1: What to Include?



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PCA Step 2: Centred Log Ratio Transform

Let's leave that for now



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PCA Step 3: Calculate PCA



- Eigenvector: How much X plus how much Y, plus (What direction?)
- 2. Eigenvalue: What proportion of overall variation (How long?)
- 3. Scaled Eigenvector: Scaled by eigenvalue. Most useful output of all. Sum of squares for each variable sums to 1.
- 4. PC Score: Principal component score for individual samples



PCA Step 4: Interpret Ranked Scaled Eigenvectors

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Two Orogenic Gold Signatures



Examples of Gold 1 and Gold 2 ore element signatures on OSNACA Enrichment Diagrams

Note: Co, Re, Pd, Pt, In, Tl, U assays not provided



We have rapidly assessed data for 80-odd RC samples from an orogenic gold project and have the following leads to follow up:

- Mafic, felsic and sedimentary host rock signatures have been defined
- Two different styles of gold mineralisation have been identified, one "oxidised", the other "reduced". Should we target where these two systems meet?
- White mica alteration may also have been defined and requires follow-up





Scavenging

1	1 1A 1 Hydrogen 1.01	2 2A					k	(ey					13 3A	14 4A	15 5A	16 6A	17 7A	18 8A 2 He Helium 4.00
2	3 Li Lithium 6.94	4 Be Beryllium 9.01	11 Atomic number Na Element symbol Sodium Element name											6 C Carbon 12.01	Nitrogen	0 Oxygen 16.00	9 F Fluorine 19.00	10 Ne Neon 20.18
3	11 Na Sodium 22.99	12 Mg Magnesium 24.31	3 3B	4 4B	5 5B	6 6B	2 Ave 7 7B	erage aton 8	nic mass* 9 —8B—	10	11 1B	12 2B	13 Al Aluminum 26.98	14 Si Silicon 28.09	15 P Phosphorus 30,97	16 S Sulfur 32.07	17 Cl Chlorine 35.45	18 Ar Argon 39.95
4	19 K Potassium 39.10	20 Ca Calcium 40.08	21 Sc Scandium 44.96	22 Ti Titanium 47.87	23 V Vanadium 50.94	24 Cr Chromium 52.00	25 Mn Manganes 54.94	26 Fe Iron 55.85	27 Co Cobalt 58.93	28 Ni Nickel 58.69	29 Cu Copper 63.55	30 Zn ^{Zinc} 65.39	31 Ga Gallium 69.72	32 Ge Germanium 72.61	33 As Arsenic 74.92	34 Se Selenium 78.96	35 Br Bromine 79.90	36 Kr Krypton 83.80
5	37 Rb Rubidium 85.47	38 Sr Strontium 87.62	39 Y Yttrium 88.91	40 Zr Zirconium 91.22	41 Nb Niobium 92.91	42 Mo Molybdenum 95.94	43 Tc Technetium (98)	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.42	47 Ag Silver 107.87	48 Cd Cadmium 112.41	49 In Indium 114,92	50 Sn ^{Tin} 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.60	53 lodine 126.90	54 Xe Xenon 131.29
6	55 Cs Cesium 132.91	56 Ba Barium 137,33	57 La Lanthanum 138.91	72 Hf Hafnium 178.49	73 Ta Tantalum 180.95	74 W Tungsten 183.84	75 Re Rhenium 186.21	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au _{Gold} 196.97	80 Hg Mercury 200.59	81 TI Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)
7	87 Fr Francium (223)	88 Ra Radium (226)	89 Ac Actinium (227)	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (266)	107 Bh Bohrium (264)	108 Hs Hassium (269)	109 Mt Meitnerium (268)									
* If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the most stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, then it refers to the atomic mass of the stable isotope. * If this number is in parentheses, the atomic mass of the stable isotope. * If this number is in parentheses, the atomic mass of the stable isotope. * If this number is in parentheses, the atomic mass of the stable isotope. * If this number is in parentheses, the atomic mass of the stable isotope. * If this number is in pare														71 Lu Lutetium 174.97 103 Lr Lawrencium (262)				

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Scale Dependence

Exploration geologists want to isolate metal associations related to mineralisation from everything else. They vary according to scale.

- If detectable in a regional dataset, a mineralisation signal will feature on a lower order PC (e.g., PC5)
- A single point (or maybe a few) will not define a metal association in PCA at all.
 You must ALSO look carefully at probability plots.
- However, within a deposit, a metal signature will feature on PC1



Additive Indices



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Do <u>NOT</u> use raw values: See also Weighted Sum function in ioGAS • •

The use of multielement geochemistry to define mineralisation signatures isolates mineralisation from competing processes like regolith and lithology

You should always follow up a Au anomaly with pathfinder support ahead of a Au-only anomaly

The use of multielement geochemistry helps to eliminate false positive and provides more confidence to follow up subtle anomalies that are related to mineralisation

Target ranking is greatly improved







In Mineral Exploration there are THREE main things that whole-rock geochemistry can help us with

- 1. Lithology
- 2. Alteration
- 3. Metal Signatures
- 1. is for a more accurate stratigraphic framework \rightarrow better structure
- 2. is for mapping hydrothermal fluid flow \rightarrow better predicts deposit sites

3. is for more reliably identifying mineralisation, and having found it, understanding the range of signatures present



